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Sweet 16 Chemistry Gases Tournament Answer Sheet

Sweet 16 Chemistry of Gases Tournament? First round: Predict the Gaseous product. Second round: Predict the one that dissolves in water to give an acidic solution. Semis: The gas that is more dense. Finals: The winner is a greenhouse gas produced in volcanic emissions. First round: Sodium Carbonate vs. Hydrochloric acid...

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1.) Na_2CO_3 vs. HCl 2.) Zn vs. HCl 3.) H_2O_2 vs. Yeast (catalyst) 4.) Na_2SO_3 vs. HCl 5.) Cu vs. HNO_3 (concentrated) 6.) CaC_2 vs. H_2O 7.) NaOCl vs. HCl 8.) NH_4NO_2 vs. Heat First round ----- Predict the gaseous product. 2nd round- The winner dissolves in H_2O to give an acidic solution. 3rd round (semis) - The gas that is more dense (g/L) wins. Finals - The winner is a greenhouse gas produced in ...

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First round ----- Predict the gaseous product. 1.) Na_2CO_3 vs. HCl 2.) Zn vs. HCl 3.) H_2O_2 vs. Yeast (catalyst) 4.) Na_2SO_3 vs. HCl 5.) Cu vs. HNO_3 (concentrated) 6.) CaC_2 vs. H_2O 7.) NaOCl vs. HCl 8.) NH_4NO_2 vs. Heat 2nd round- The winner dissolves in H_2O to give an acidic solution. 3rd round (semis) - The gas that is more dense (g/L) wins. Finals - The winner is a greenhouse gas produced in ...

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tournament. • First round: Predict the name and formula of the gaseous product obtained in a chemical reaction involving the reactants listed in the first round.

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